**Name\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_Period\_\_\_\_\_\_Date\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_**

**Net Ionic Reaction Practice #2**

(a) write the molecular equation with symbols, (c) write the net ionic equation for each of the following:

1. Silver nitrate + potassium chromate
2. Ammonium chloride + cobalt (II) sulfate
3. Lithium hydroxide + sodium chromate
4. Zinc acetate + cesium hydroxide
5. Ammonium sulfide + Lead (II) nitrate
6. Iron (III) sulfate + barium iodide
7. Chromium (III) bromide + sodium nitrate
8. Rubidium phosphate + titanium (IV) nitrate
9. Ammonium carbonate + nickel (II) chloride
10. Tin (IV) nitrate + potassium sulfite